

Course Content Form

PREFIX CHM 121N

COURSE NAME: Chemistry and Society

Instructor: Lucinda Begay

Campus: Main

Semester Dates: Summer 2018 Class Days/Times: MTWTH/3:45-

5:45pm

Credit Hours: 4

Lecture Periods: 3

Lab Periods: 3

Instructor Information: Phone (message only): 623-6175

Name: Lucinda Begay E-mail: lbegay@tocc.edu

Office location: N/A

Office hours: Appointments before or after

class

Prerequisites: MAT 92 or placement in MAT 142H

Course Description:

This course covers the foundations of chemistry and its relationship to everyday experiences. Skills such as conversions and problem-solving techniques are also covered. Students will be introduced to laboratory techniques and participate in experimental activities that promote the development of basic and advanced science-process skills.

Information: (Please include any information specific to this course.)

N is the integrated version of the course with the lecture and lab taught simultaneously

Student Learning Outcomes:

Upon completion of the course, the student will be able to do the following:

- 1. Differentiate between physical and chemical properties and changes, and between elements, compounds, and mixtures.
- 2. Discuss current science problems and potential problems of the future.

- 3. Use appropriate chemical terminology and conventions to interpret symbols and formulas, balance chemical
- equations, name and write formulas for common inorganic compounds, and identify various types of chemical reactions.
- 4. Perform calculations involving the metric system, scientific notation, the mole concept, and concentration terms including molarity, percent, ppm, ppb and dimensional analysis.
- 5. Use periodic table to identify metals, nonmetals, and metalloids, and list general physical properties for each category including Lewis dot structure.
- 6. Identify and properly use lab equipment such as flasks, balances, beakers etc.

Course Outline:

- I. Classification and Structure of Matter
 - A. Atomic, ionic, or molecular nature of all material
 - B. Atomic structure
 - C. Electronic arrangement
- II. Radioactivity
 - A. Nuclear decay, radiation types
 - B. Power source
 - C. Societal implications, current usage
- III. Compound Formation from Elements
 - A. The Periodic Table of Elements
 - B. Metals and non-metals; their valence electrons, and periodic perspective
 - C. Atomic elements to ionic or molecular compounds; electron transfer and sharing
- IV. Electron Transfer: Electricity, Oxidation and Reduction
 - A. The electrochemical cell
 - B. Construction of batteries
 - C. Corrosion, rust, and bleaching
- V. Acids, Bases and Salts
 - A. Definition and reaction with indicator dyes
 - B. pH Scale
 - C. Common acids and bases; and their relative strengths
 - D. "Acid Rain" or other environmental topics related to pH
- VI. Chemical Quantities
 - A. The mole concept
 - B. Calculation of molar mass of pure substances
 - C. Interconversion of mass, mole and number of particles
 - D. Molar solutions
- VII. Stoichiometry
 - A. The balanced equation
 - B. Equation molar and mass relationships

VIII. The Liquid State

- A. Water and its unique properties
- B. Solutions and solubility
- C. Contaminants and purification

IX. The Gas State

- A. Comparison to solid, liquid states: particle density
- B. Effects of temperature and pressure on volume
- C. The atmosphere and atmospheric pressure
- D. Modeling the gaseous state

X. Special Topics

- A. Traditional Knowledge of Chemistry
- B. Lab techniques: use of flasks, balances and other classroom
- C. appropriate activities
- D. Other special topics selected by instructor as relevant

Evaluation and Grading & Assignments:

Exams: There are 4 exams during the course of the semester.

	Total Points
400 (4 @ 100 pts)	40%
80 pts	25%
160 pts	25%
60 pts	10%
700	100
	80 pts 160 pts 60 pts

Himdag Cultural Component:

Course will explore existing elements, materials and environmental issues where this applies to culture chemical thinking to continue improving or restoring components of Himdag. Students will also formulate a personal ethic regarding the use of chemical thinking in science teaching and research, incorporating perspectives from Western science and TOCC Core values.

CHM 121IN—Chemistry

TENTATIVE SCHEDULE SUMMER 2018 BEGAY

Date	Assigned	Topic	Assignments, Labs, Class
	Reading		Activities
5/29-	Ch. 1-2	Intro to Class, syllabus, chemical	Lab 1:
5/31		safety,	HW1
		Measurement	Quiz 1
6/4-6/7	Ch. 3-4	SI Unit, Matter	Lab 2:
			HW2
			Quiz 2
6/11-	Ch. 5	Atoms and Molecules	Lab 3
6/14		Review/EXAM 1	HW 3
			Quiz 3
6/18-	Ch. 6-7	Chemical Periodicity	Lab 4
6/21		Chemical Bond	HW 4
			Quiz 4
6/25-	Ch. 8	Review/ EXAM 2	Lab 5
6/28		Chemical Nomenclature	HW 5
			Quiz 5
7/2-7/5	Ch. 9-10	Chemical Calculations	Lab 6
			HW 6
			Quiz 6
7/9-	Chemical	Review/EXAM 3	Lab 7
7/12	Application		HW 7
			Quiz 7
7/16-	Chemical	Chemical Equations	Lab 8
7/19	Application	-	HW 8
			Quiz 8
7/23-		Review/Final Exam/Lab Exam	
7/24			

DISCLAIMER: This syllabus is designed to evolve and change throughout the semester based on class progress and interests. You will be notified of any changes as they occur.

Acknowledgment of Receipt of Syllabus

Please read, sign and return the following acknowledgment to me in class, *or* return to me at the following address:

Lucinda Begay Tohono O'odham Community College P.O. Box 3129 Sells, AZ 85634

	I have received my CHM 121IN syllabus (including course obschedule) and have read and understood all the enclosed ma				
	I prefer that the instructor not call or contact me by phone anytime during the semester.				
Му	y reason(s) for taking this course:				
Му	y background in this area includes:				
	${ m l}$ I would like to be contacted by the instructor regarding the fo	llowing concerns:			
	Print Name Clearly Here	Sign Name Here			
	Student ID Number	Telephone Number			
	Current Mailing Address/City/State/Zip	E-mail Address			